

**The Membrane Treatment Solution** 

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Phosphorus is ubiquitous in wastewaters. When used appropriately, it is a major resource, e.g. for the agricultural industry.

Uncontrolled release of phosphorus, however, could lead to eutrophication, which significantly affects the river eco-system.

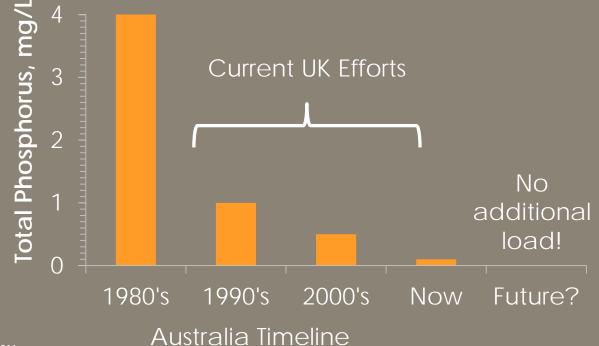
#### Outline of Presentation

1. The phosphorous limits and typical removal technologies

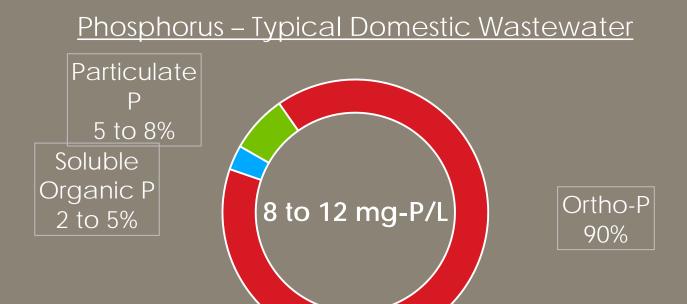
2. The Googong Project - Phosphorous removal using membranes

3. Questions

## Effluent P limit is going down....



#### Phosphorus Load



### Phosphorous Removal

Phosphorous has to leave the system as particulate-bound compounds

Two main removal mechanisms:

- 1. Chemical precipitation
- Co precipitation (1-2 mg/L)
- Multipoint / tertiary (0.1 0.5 mg/L)
- 2. Enhanced (excess) biological phosphorous removal

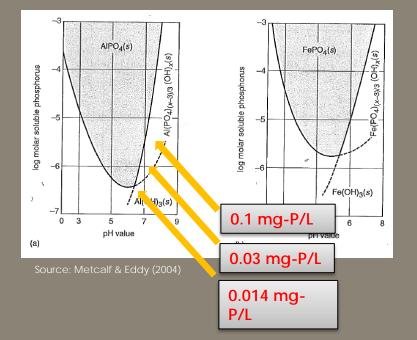






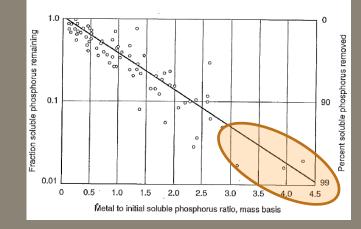
### Choice of Coagulants

Ferric or ferrous (as chloride or sulphate) Alum (sulphate), or ACH



Phosphorus reduction, %	Mole ratio, Al:P		
	Range	Typical	
75	1.25:1-1.5:1	1.4:1	
85	1.6:1-1.9:1	1.7:1	
95	2.1:1-2.6:1	2.3:1	

<sup>a</sup>Developed in part from U.S. EPA (1976).

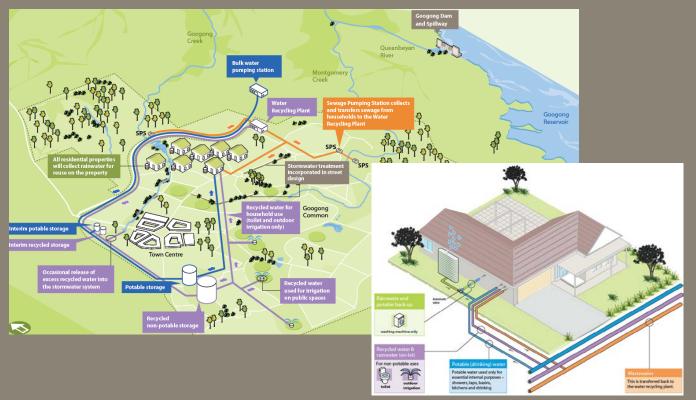


## The Googong Story

In early 2000's, a new town was being planned for development near Canberra.



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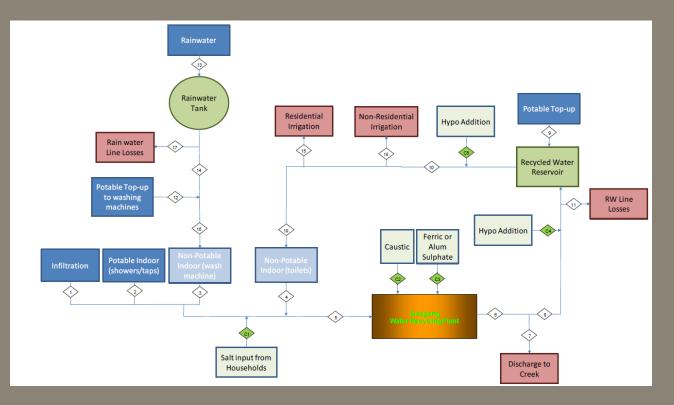


#### The Phosphorous & TDS Challenges!

- Target 0.3 mg/L 50% P discharge to Molonglo River
- Water Recycling semi-closed water loop → TDS accumulation
- Important to consider both phosphorous and TDS at the same time.
- Additional project challenges:
- 1. Need for membrane filtration
- 2. Land is premium!



#### Water & TDS Modelling



## **TDS Balance Key Outcomes**

Phosphorous precipitating chemicals (alum and/or ferric) are the <u>second highest</u> contributor to TDS in water cycle.

- It is important to minimize chemical dose
- Overdose will lead to depressed pH, meaning more caustic dose, which also increases TDS
- Biological phosphorous removal could improve efficiency of chemical uses

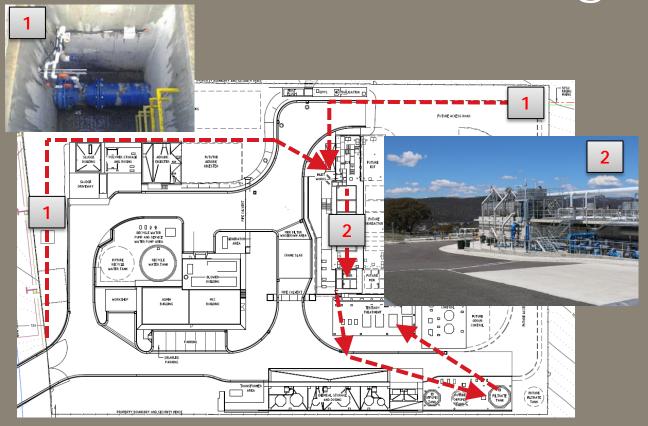




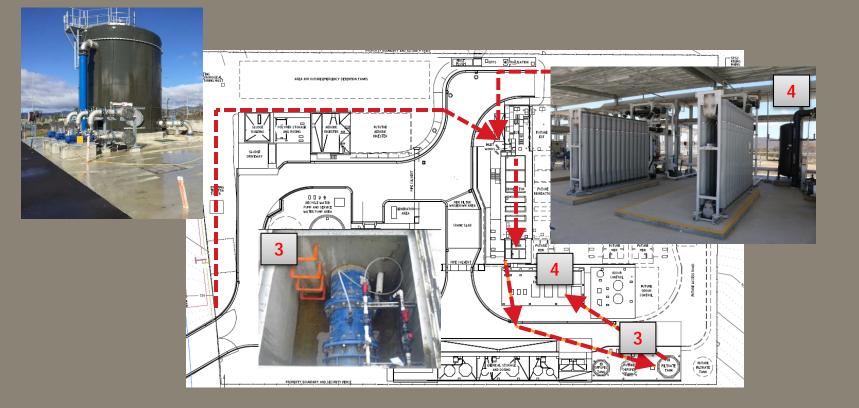
### P Removal Process Design

- 1. Multi-point dosing + separation to minimize chemical dosage (MBR followed by pressure UF)
- 2. Recycle tertiary precipitates to bioreactor to maximize chemical efficiency
- 3. Managing sludge return liquor
- 4. Ability to operate bioreactor in enhanced biological phosphorous removal mode

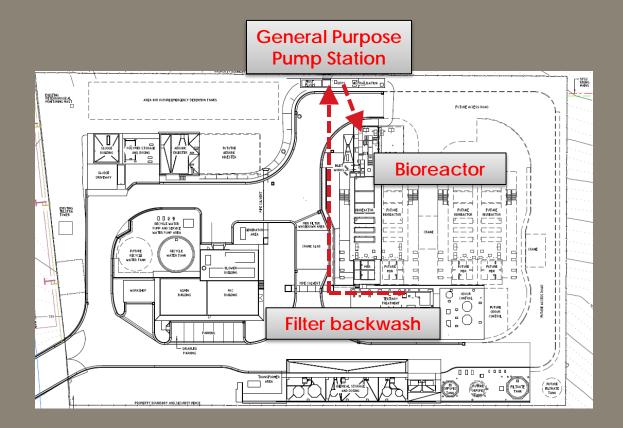
#### Process Flow & Ferric Dosing



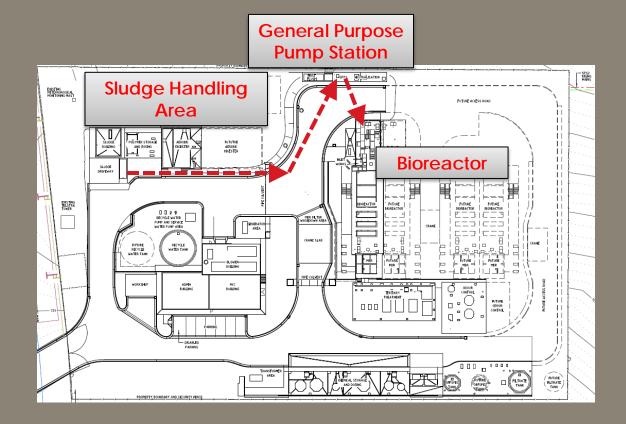
#### Tertiary Phosphorus Removal



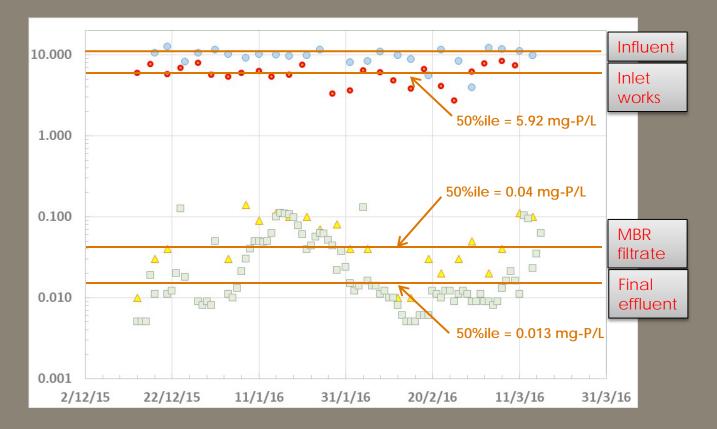
#### Reuse of Residual Alum Precipitate



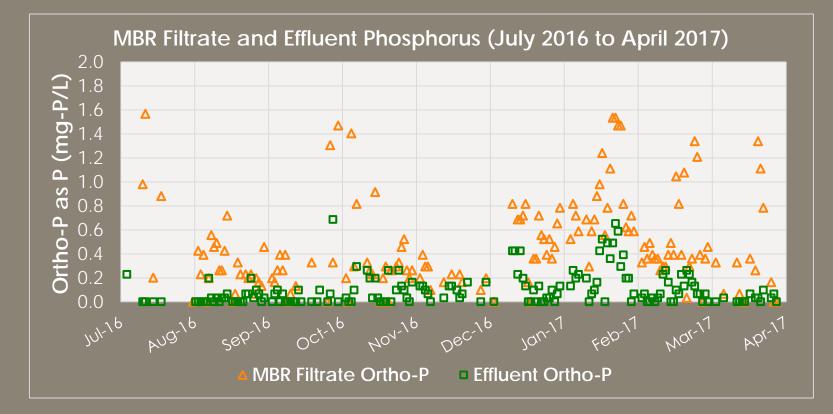
## Sludge Liquor Management



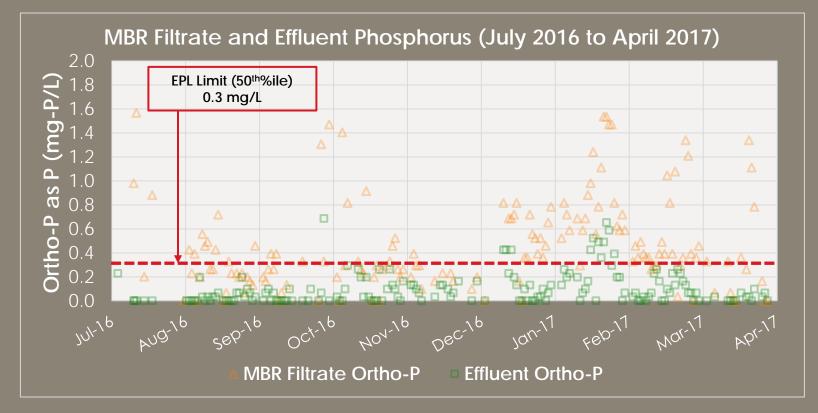
### Commissioning Outcomes



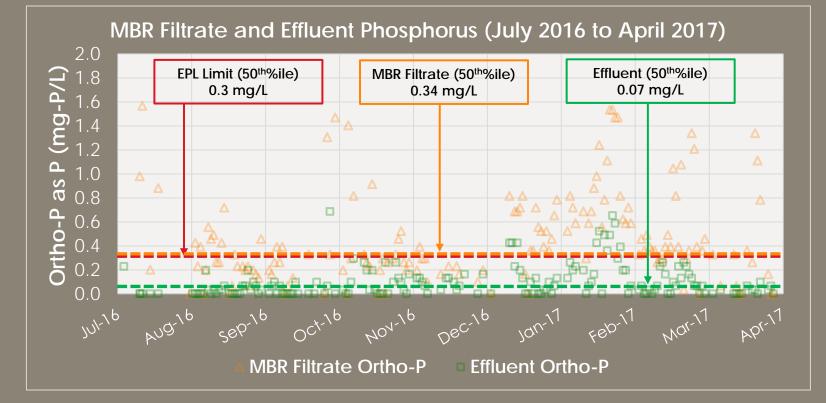
#### **Operating Outcomes - Phosphorus**



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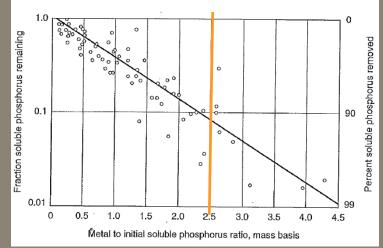


#### What about Chemical Consumption?

#### Ferric usage

- First dose (sewer main):
  - 3.1 mg-P/L removed @ 1.5 Fe : P molar ratio
  - Ferric also suppresses H<sub>2</sub>S gas
- Second dose (bioreactor):
  - 6.6 mg-P/L removed
  - Fe : P molar ratio = 2.1 or 3.1 (allowing for effect of alum dose return)

Overall Fe+AI : P molar ratio = 2.5, with 99.4% removal



#### What about TDS?

TDS impacts ~ 106 mg/L SO $_4$  attributable to P precipitating chemicals

Process	Influent (mg/L)	Effluent (mg/L)	Difference (mg/L)
P removal	31 (PO4)	106 (SO4)	+75
N removal	64 (NH3)	4 (NO3)	-60
Disinfection & membrane cleaning		17 (Na) 12 (Cl)	+29
Bicarbonate	311 (alkalinity)	63 (alkalinity)	-151
Others (Mg, Ca)	26	28	+2
		Total:	-105

#### Summary & Lessons Learnt

Globally, phosphorous limit is trending down

Technology is now available to achieve near-zero P

Lessons learned from the Googong project:

- Multi-point dosing (with ferric & alum) is key to achieving chemical efficiency. It also increases process robustness,
  - Process / equipment breakdown
  - Supply of chemicals
- Managing shock loads well gives consistent performance
- Membrane removes fine or colloidal phosphorous, allowing very low limit to be achieved

## Ongoing & Future Works

#### Flow monitoring works

- Understand amount of water recycled in the loop to clarify TDS modelling works

#### Optimize chemical additions

- Further reduce ferric and alum doses
- Using citric acid (instead of sulphuric acid) for membrane cleaning

# Thank you!

#### Questions?

