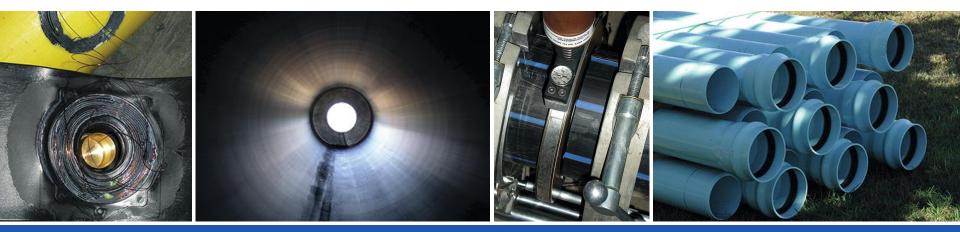


#### **Disinfection Residual Effects On Water Distribution Systems Materials**

Robert J LeHunt LeHunt & Associates Pty Ltd

Water New Zealand Conference & Expo 2019

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# **Residual Disinfectants Cause Deterioration To Network Materials**

Complex Reactions
Depend on Type & Concentrations



# Australian and New Zealand Drinking Water Guidelines

Disinfectant Residual Levels Based on Aesthetic and Health Related Issues



# **Typical Guideline Levels**

Free Allowable Chlorine 0.6 mg/L

- Typical System Levels 0.2 1.5 mg/L
- Maximum Allowable Chlorine 5 mg/L
- Maximum Allowable Chloramine 3 mg/L
- Typical Chlorine Dioxide Levels 1 2.5 mg/L
- Target Levels pH 6.5 8.5



# **Residual Disinfection Types**

- Chlorine
- Chloramine
- Chlorine Dioxide



## Chlorination (Gas/Liquid addition)

## $CI_2 + H_2O \rightarrow HOCI + H^+ + CI^-$

HOCI (Hypochlorous acid)

- Weak acid, powerful oxidising agent
- Primary disinfecting agent



# **Chlorination** (Gas/Liquid addition) **Further dissociation** $HOCI \leftrightarrow H^+ + OCI^-$

- H<sup>+</sup> Important Hydrogen ions not drop too low hence pH not too high  $\leq$  7.5
- Water should be in range 7.0 8.0



# **Chloramination**

### $\mathsf{HOCI} + \mathsf{NH}_3 \leftrightarrow \mathsf{NH}_2\mathsf{CI} + \mathsf{H}_2\mathsf{O}$

#### NH<sub>2</sub>Cl (Monochloramine)

• Controlled by pH and Chlorine:Ammonia ratio



# Chlorine Dioxide (Most common process)

## $2\text{NaClO}_2 + \text{Cl}_2 \rightarrow \textbf{2ClO}_2 + 2\text{NaCl}$

#### ClO<sub>2</sub> (Chlorine Dioxide)

- Effective over wider water pH range
- Decomposed rapidly by sunlight and UV light



# **Chlorine Dioxide**

Reacts with Hydrogen Peroxide (HOOH)

## $2\text{CIO}_2 + \text{HOOH} \rightarrow 2\text{HCIO}_2 + \text{O}_2 \quad (1)$

Probably reacts with polymeric hydroperoxides (ROOH)

## $CIO_2 + ROOH \rightarrow RO_2 + HCIO_2$ (2)

- Chain branch reaction as is thermal degradation
- Proposed accelerates polymeric oxidation rate



# **Disinfection Aggression Differences**

#### **Chlorine Dioxide** Chloramine Chlorine

- Chlorine Dioxide most aggressive
- Disinfection method change can result in existing material degradation



| Elastomer                 | Chlorine pH 8.5                             | Chloramine pH 8.5                        |
|---------------------------|---|--|
| Neoprene                  | Minor crack<br>Surface distort              | Moderate crack<br>Minor<br>embrittlement |
| Nitrile<br>(Sulphur cure) | Minor crack<br>Surface distort              | Minor crack                              |
| EPDM<br>(Sulphur cure)    | Minor crack                                 | Heavy crack<br>Minor<br>embrittlement    |
| Natural Rubber            | Moderate crack<br>Moderate<br>embrittlement | Heavy crack<br>Moderate<br>embrittlement |

• USA changes from chlorine to chloramine resulted in number of premature elastomer failures (19)



# Secondary Disinfection Effects Copper (Cu)

- Copper corrosion complex, due to more than one influence
- HOCI primary oxidant causing Cu<sub>2</sub>O corrosion scale on pipe surface
- CIO<sub>2</sub> and NH<sub>2</sub>CI can cause dissolution of scale releasing Cu<sup>+</sup> ions into the water
- Cause accelerated failures in other materials (PPr) in hot water systems
- Recommendations that  $CIO_2$  not be used as a disinfectant (17)

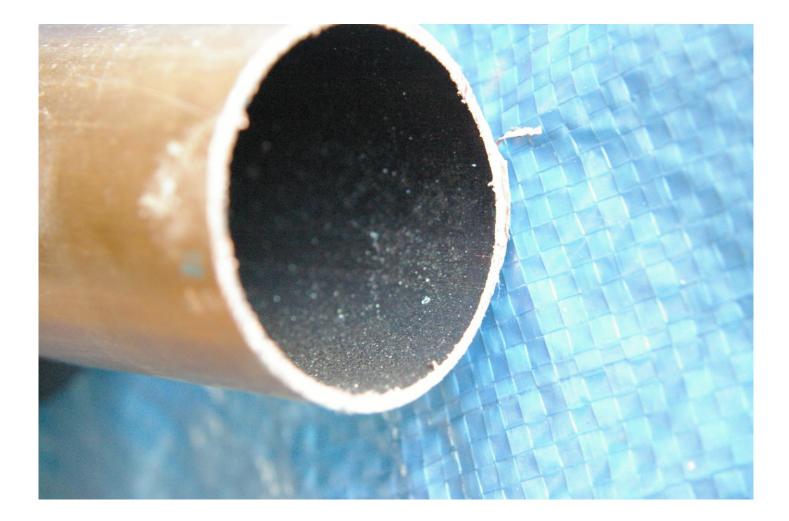


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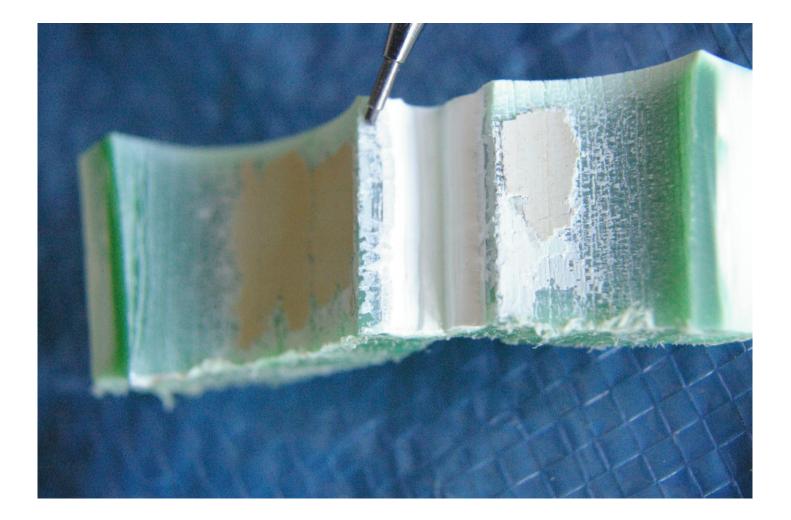


#### **Copper Internal Corrosion Layer**



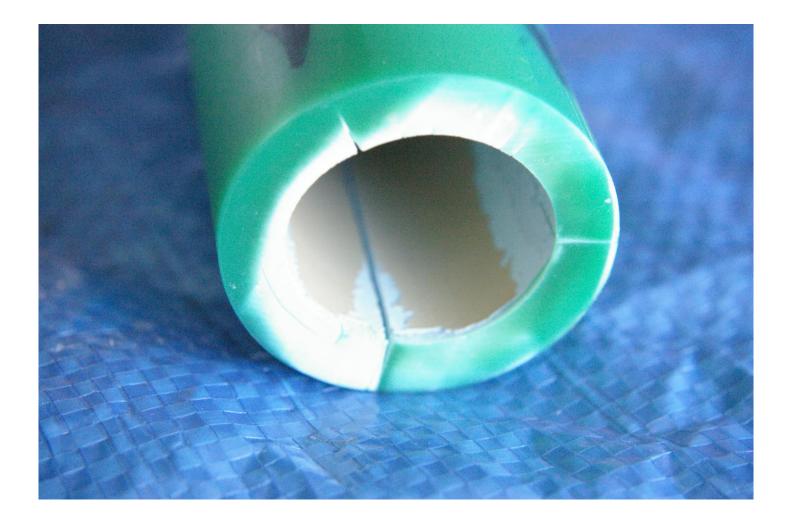


#### **PPr Oxidation Crazing and Partial Initial Fracture – High Cl<sub>2</sub>**



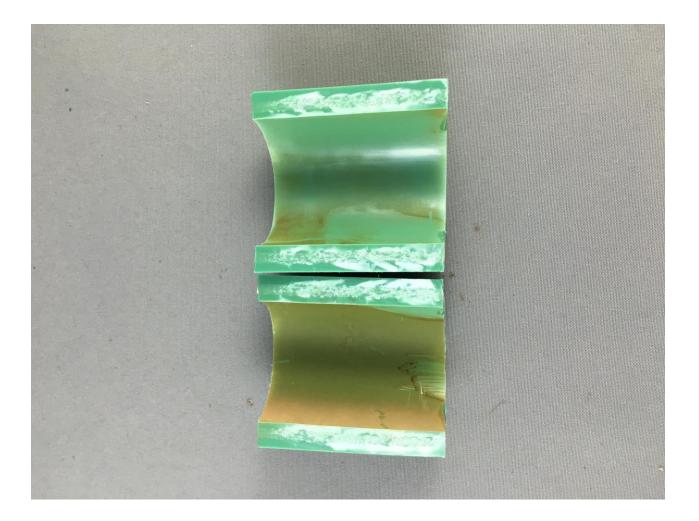


#### **PPr Oxidation Crazing Crush Testing Fracture Propagation**





#### **PPr High Cu Content Film – No Oxidation Cracking**









- Commonly used water network material
- HDPE (PE63  $\rightarrow$  PE80  $\rightarrow$  PE100) Practical use 1955 2019 ~ 60 years

### Arrhenius Reaction Rate = $Ae^{-E/RT}$

- E Activation energy of reaction (50- 150 kJ/mol/K)
- **T Absolute Temperature**
- R Universal Gas Constant 8.3 J/mol/K

Critical Time  $I_n$  (critical time) = A/T + E

**Predicts resistance to thermal oxidation** 

UL (Underwriters Laboratory USA) HDPE 50° C Air 100,000 hours (about 11.4 years)

Rerate to 15° C material temperature Lifetime → 119.7 years



- Oxidation damage shows up as fine cracking/crazing on inside diameter of pipe
- Propagation of fine cracks through the pipe wall until burst occurs
- Shows up as brittle rupture several small cracks
- ISO 9080 long term stress/time regression curve point Stage III onset







- Sporadic failures reported in water services
- France Change from Chlorine to Chlorine Dioxide Disinfection
- North Africa High temperature, Uncertain pipe quality, Uncertain disinfection levels

 North West Australia/Far North Queensland – High soil temperatures, High water temperatures, Uncertain disinfection levels, Uncertain installation practices, Small diameter thin wall pipes

#### All These Factors Reduce Nominal Pipe Lifetimes



#### **Polyethylene (PE) Soil Temperatures**

| Location                                    | Soil Temperature °C<br>0.3 – 1.0 metre depth<br>Annual average BOM |
|---|--|
| Adelaide/Perth/Melbourne<br>Sydney/Brisbane | < 21   |
| Mt Isa/Katherine/Tennant Ck                 | 27 – 30  |
| Port Headland*                              | 30 – 33  |

- Reported failures (all causes) > 20 bursts(leaks)/100 Km pipe\*
- No reported failures

\* CEED Seminar 2017

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| Location         | Soil Temperature<br>100 mm NIWA<br>Mean monthly °C | Soil Temperature<br>100 mm NIWA<br>Year average °C |
|------------------|--|--|
| Dunedin          | 10 – 20  | 10.0   |
| Christchurch     | 4 – 18   | 10.6   |
| Palmerston North | 7 – 18   | 12.8   |
| Auckland         | 10 - 20  | 15.1   |
| Kaitia           | 11 - 20  | 15.2   |



## Anticipated that the 2020 Revision of AS/NZS 2033 – Installation of PE Pipeline Systems; will include Temperature/Lifetime values for PE80 and PE 100 materials



## **Disinfection/Oxidation Stabilisation**

- 2 types of anti-oxidation stabilisation
- \* High temperature (200 °C) for short term processing
- \* Lower temperature for long term leaching/depletion during service lifetime



## **Disinfection Stabilisation USA Approach**

Immerse PE Pipe Specimens in recirculating water at 20°C with replenished Chlorine level at 4 mg/l and ORP of 800 mV.

### ASTM F 2263:2014 Method of Test

Establishes rupture stress/time points to categorise materials



## Test results classified against ASTM D3350:2014 – Standard Specification for PE Plastics Pipes and Fittings Materials

Class CC1 Base resin (existing PE100 AS/NZS 4131)

**Class CC2 Higher oxidative resistance** 

**Class CC3 Highest oxidative resistance** 



Designed for small diameter/thin wall pipes operating in high temperature/high disinfection (Chlorine/Chloramine) content applications

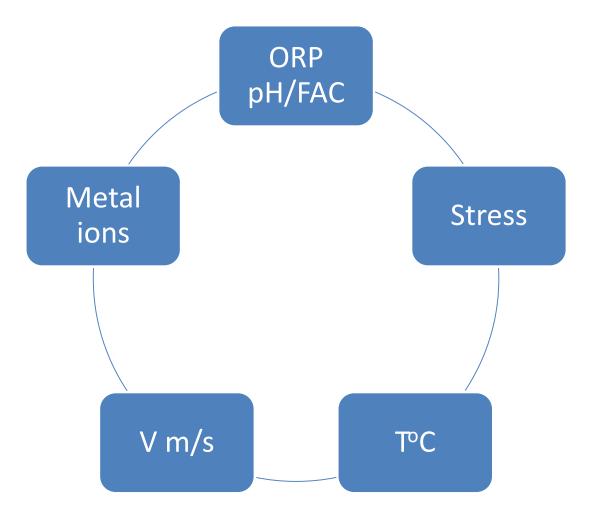
Debate exists as to Chlorine Dioxide application due to possible damage to polymer chains

Recommendations PIPA/PPI USA to not use Chlorine Dioxide as a disinfectant\*

\*PIPA POP 018, PPI TN44 - 2015



#### Oxidative performance needs combination of all inputs





# Thank you

## Welcome any questions Will be at Stand 165 over Expo